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PLAY. single-
displacement.

the replacement
of hydrogen from
water by sodium

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is an example of
a _____
reaction.

activation
energy. in order
for a chemical
reaction to take
place, the
particles
involved must
collide with a
sufficient
amount of _____

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1) calculate moles of each element (if given mass) 2) divide each mole quantity by the smallest mole value from step 1. 3) use whole number values

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from step 2 as the subscripts to the original equation to write the empirical formula. Percent Composition.
-the relative amounts of each element in a compound.

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Study Guide.
Molecule. Ionic
Bond. Nonpolar
Covalent Bond.
Polyatomic Ions.
unit of matter
that results
when 2 or more

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atoms are joined by... A bond resulting from the electrostatic attraction of a cation... A bond characterized by an even distribution of charge; electr...

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Multiple Choice
24) What is the

amount of heat
required to

raise the

temperature of

200.0 g of

aluminum by

10°C? cal

(specific heat

of aluminum =

0.21 J/g x °C

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24) A) 4200 cal
B) 420 cal C) 42
000 cal D) 420
000 cal E) none
of the above

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the choice that
best completes
the statement or
answers the
question. _____

1. How many
moles of
tungsten atoms
are in $4.8 \times$
 10^{25} atoms of
tungsten? a. 8.0
 $\times 10^2$ moles c.
 1.3×10^{-1} moles

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b. 8.0×10^1
moles d. $1.3 \times$
 10^{-2} moles _____
2.

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History of
chemistry before
1701 Chemistry
had a bad
reputation
because of

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alchemy, which was a pseudoscience that claimed to convert cheap metals into gold. Alchemy led to the discovery of many elements.
1500 - George Bauer discovered metallurgy, extracting pure

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metal out of a
mix 1600s -
Robert Boyle
studied the
behavior of
gases and is
responsible for
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elements ...

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